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#### **Key indicators**

Single-crystal X-ray study T = 120 KMean  $\sigma$ (C–C) = 0.002 Å R factor = 0.036 wR factor = 0.108 Data-to-parameter ratio = 15.2

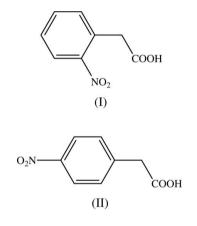
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved Molecules of the title compound,  $C_8H_7NO_4$ , are linked into centrosymmetric  $R_2^2(8)$  dimers by paired  $O-H\cdots O$  hydrogen bonds, and these dimers are linked by two  $C-H\cdots O$  hydrogen bonds into sheets of  $R_2^2(8)$  and  $R_4^4(18)$  rings.

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### Comment

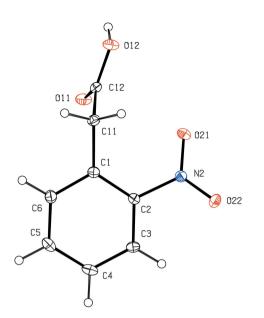
As part of our investigations of compounds containing nitro and carboxylic acid groups (Glidewell *et al.*, 2003*a*,*b*, 2004, 2006; Wardell *et al.*, 2005), we now report the molecular and supramolecular structure of 2-nitrophenylacetic acid, (I) (Fig. 1).

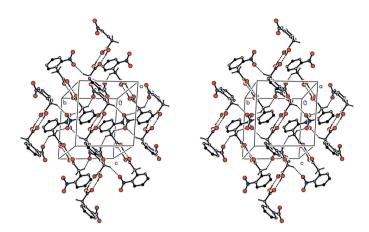


The plane of atoms C1/C11/C12 is almost orthogonal to the plane of the aryl ring (Fig. 1, Table 1), while the  $C-NO_2$  plane makes a dihedral angle of 30.1 (2)° with the ring.

The molecules of (I) are linked into sheets by a combination of N-H···O and C-H···O hydrogen bonds (Table 2). Paired O-H···O hydrogen bonds link the molecules into centrosymmetric  $R_2^2(8)$  (Bernstein *et al.*, 1995) dimers (Fig. 2). Two C-H···O hydrogen bonds link the dimers, so forming a (100) sheet built from  $R_2^2(8)$  and  $R_4^4(18)$  rings. The resulting net is of type (4,4) (Batten & Robson, 1998). There are no direction-specific interactions between adjacent sheets. In particular, C-H··· $\pi$ (arene) hydrogen bonds and aromatic  $\pi$ - $\pi$  stacking interactions are both absent.

The structure of the isomeric 4-nitrophenylacetic acid, (II), was reported some years ago [Cambridge Structural Database (Version of November 2005; Allen, 2002) refcode SEMTAF; Grabowski *et al.*, 1990]. The authors reported the formation of a centrosymmetric hydrogen-bonded dimer, but further aggregation of the dimers was not reported. In the event, the dimers are linked into sheets by a single aromatic  $\pi$ - $\pi$  stacking interaction (Fig. 4).



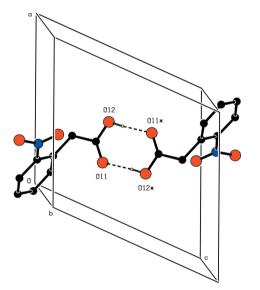


#### Figure 3

A stereoview of part of the crystal structure of (I), showing the formation of a (100) sheet of  $R_2^2(8)$  and  $R_4^4(18)$  rings. For the sake of clarity, H atoms bonded to aromatic C atoms have been omitted.

## Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.



#### Figure 2

Part of the crystal structure of (I), showing the formation of a centrosymmetric  $R_2^2(8)$  dimer. For the sake of clarity, H atoms bonded to C atoms have been omitted. Atoms marked with an asterisk (\*) are at the symmetry position (1 - x, 1 - y, 1 - z).

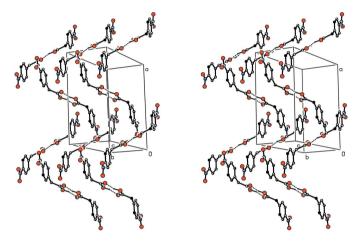
### **Experimental**

A commercial sample of (I) (Acros) was crystallized from ethanol (m.p. 412-413 K).

### Crystal data

C<sub>8</sub>H<sub>7</sub>NO<sub>4</sub>  $M_r = 181.15$ Monoclinic,  $P2_1/c$ a = 9.3182 (3) Å b = 9.4466 (2) Å c = 9.9733 (3) Å  $\beta = 114.7990 (17)^{\circ}$ V = 796.95 (4) Å<sup>3</sup>

Z = 4 $D_x = 1.510 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation  $\mu = 0.12 \text{ mm}^{-1}$ T = 120 (2) KLath, colourless  $0.52 \times 0.26 \times 0.10 \text{ mm}$ 



### Figure 4

A stereoview of part of the crystal structure of (II), showing the formation of a sheet of  $\pi$ -stacked hydrogen-bonded dimers. The original atomic coordinates (Grabowski et al., 1990) have been used. For the sake of clarity, H atoms bonded to C atoms have been omitted.

#### Data collection

Bruker Nonius KappaCCD area-	8852 measured reflecti
detector diffractometer	1829 independent refle
$\varphi$ and $\omega$ scans	1682 reflections with I
Absorption correction: multi-scan	$R_{\rm int} = 0.031$
(SADABS; Sheldrick, 2003)	$\theta_{\rm max} = 27.7^{\circ}$
$T_{\min} = 0.949, \ T_{\max} = 0.988$	

#### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.037$ wR(F<sup>2</sup>) = 0.108 S = 1.161829 reflections 120 parameters H-atom parameters constrained tions ections  $> 2\sigma(I)$ 

 $w = 1/[\sigma^2(F_{\rm o}{}^2) + (0.0444P)^2$ + 0.341P] where  $P = (F_0^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.34 \text{ e } \text{\AA}^{-3}$  $\Delta \rho_{\rm min} = -0.27 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 (Sheldrick, 1997) Extinction coefficient: 0.103 (10)

Table 1		
Selected torsion	angles	(°).

C2-C1-C11-C12	83.34 (16)	C1-C2-N2-O21	-29.64(17)
C1-C11-C12-O12	-159.51 (11)		

 Table 2

 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O12{-}H12{\cdots}O11^i$	0.84	1.83	2.6622 (14)	173
$C11-H11A\cdots O21^{ii}$	0.99	2.35	3.1758 (16)	140
$C11 - H11B \cdots O22^{iii}$	0.99	2.54	3.4398 (19)	151
				4

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (iii) -x + 1, -y + 1, -z.

All H atoms were located in a difference map and then treated as riding, with C-H distances of 0.95 Å (aromatic) or 0.99 Å (CH<sub>2</sub>), and O-H distances of 0.84 Å, and with  $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C)$  or  $1.5U_{\rm eq}(\rm O)$ .

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT*; data reduction: *DENZO* and *COLLECT*; program(s) used to solve structure: *OSCAIL* (McArdle, 2003) and *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *OSCAIL* and *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PRPKAPPA* (Ferguson, 1999).

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